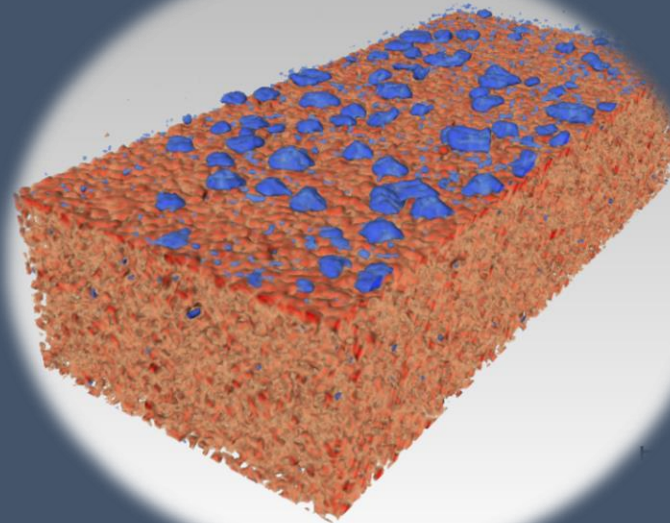




Solid State Batteries

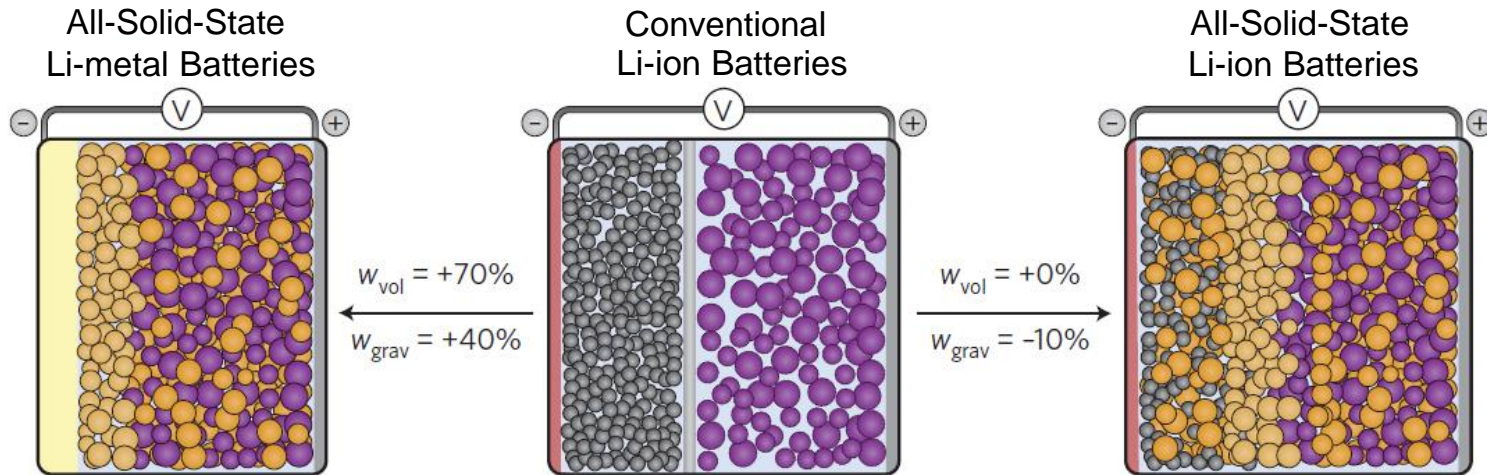
The role of Li metal voids in cell failure

Dominic Spencer Jolly
University of Oxford



THE ALL SOLID STATE LITHIUM BATTERY

from liquid to solid electrolytes



W_{vol} and W_{grav} = Volumetric and Gravimetric energy densities, respectively

Zeier, W. G. & Janek, *Nat. Energy* 1, 1–4 (2016)

Advantages:

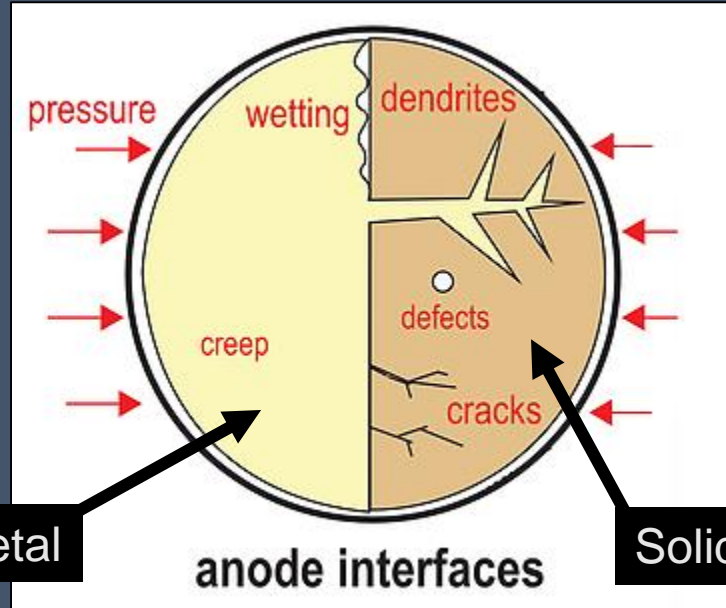
- **Safety – no flammable organic liquid electrolyte**
- **Enable use of Li metal anode - higher energy density**
- **Longer life**
- **Faster charge/discharge**

Energy Density	Current Li-ion	Projected ASSBs
Gravimetric (Wh/kg)	250	500
Volumetric (Wh/L)	700	1300

Challenges of the lithium metal anode/solid-electrolyte interface

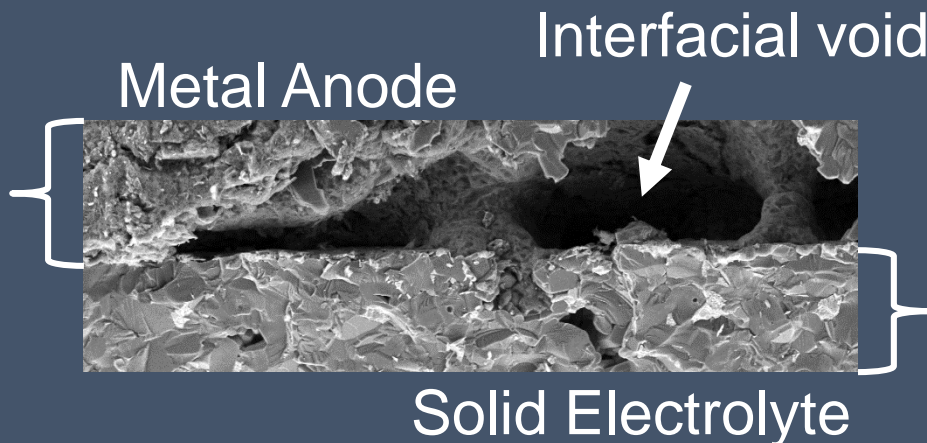
Charge

Void formation occurs on stripping

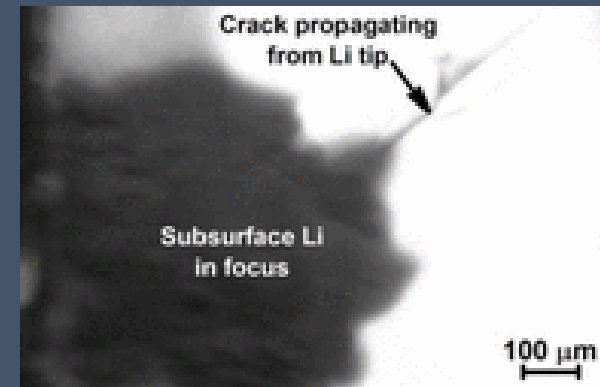


Discharge

Dendrite formation occurs on plating even if ceramic > 95% dense

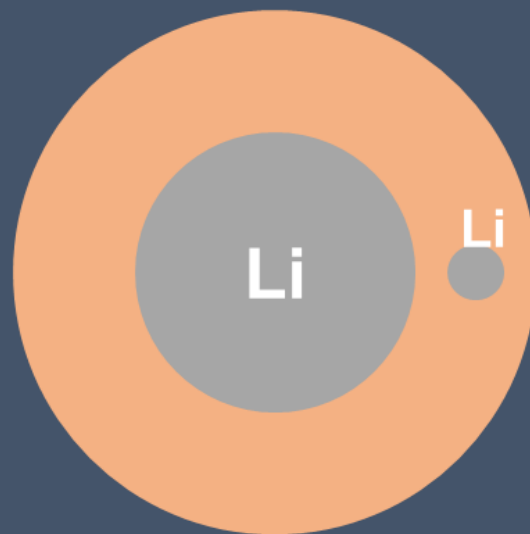
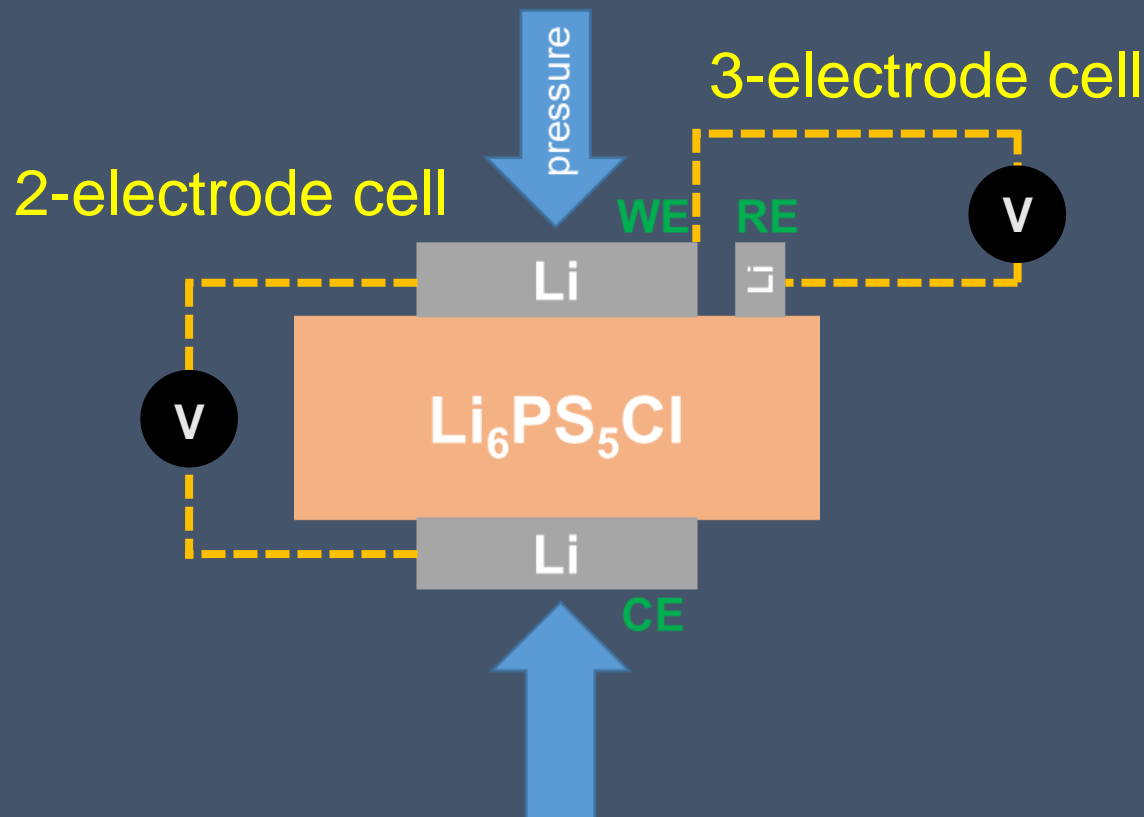


Single crystal $\text{Li}_{6.25}\text{Al}_{0.25}\text{La}_3\text{Zr}_2\text{O}_{12}$

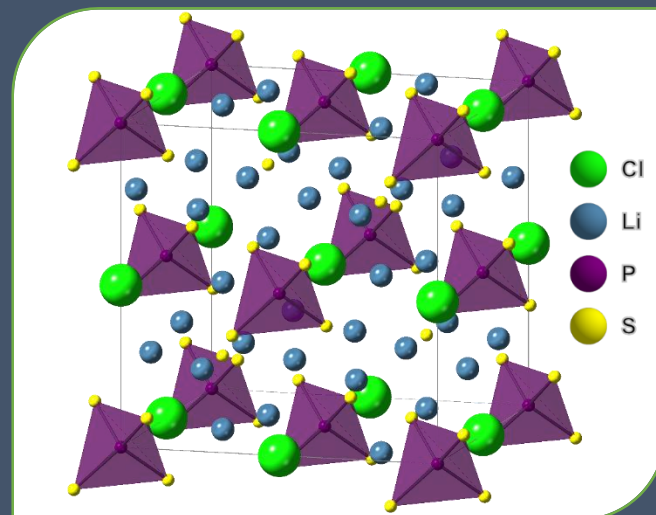


Chiang, et al. *JES*, 2018, 165, A3648.

3-electrode cell – separate stripping from plating



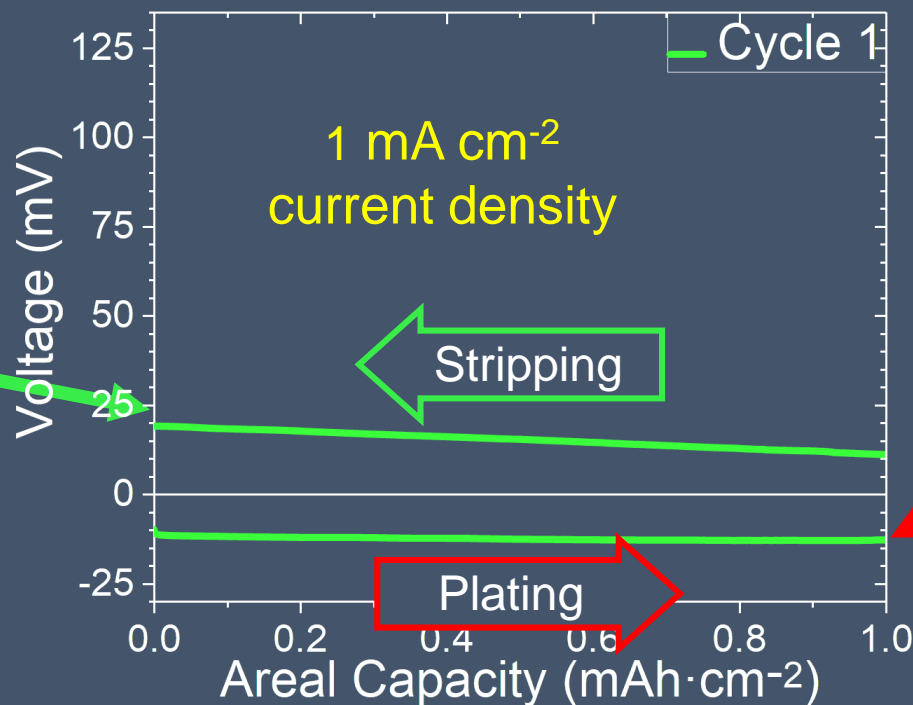
- 3-electrode cell allows separation of Li plating from stripping
- Argyrodite $\text{Li}_6\text{PS}_5\text{Cl}$ solid electrolyte:
 - conductivity $> 10^{-3} \text{ S cm}^{-1}$
 - soft
- Important to control pressure



3-electrode cycling Li/Li₆PS₅Cl 3 MPa

*Kasemchainan et al.
Nature Materials (2019)*

Small polarisation increase on stripping

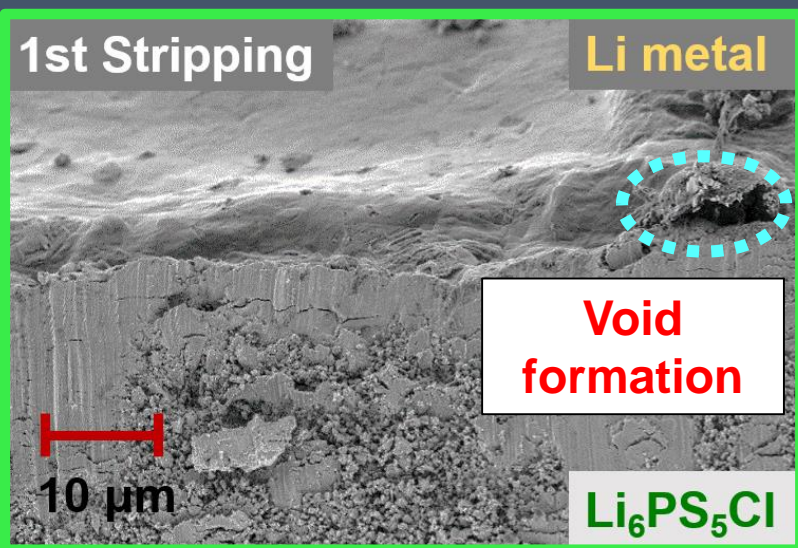


No change in polarisation on plating

Strip/plate 5 μm of Li

1st Stripping

Li metal



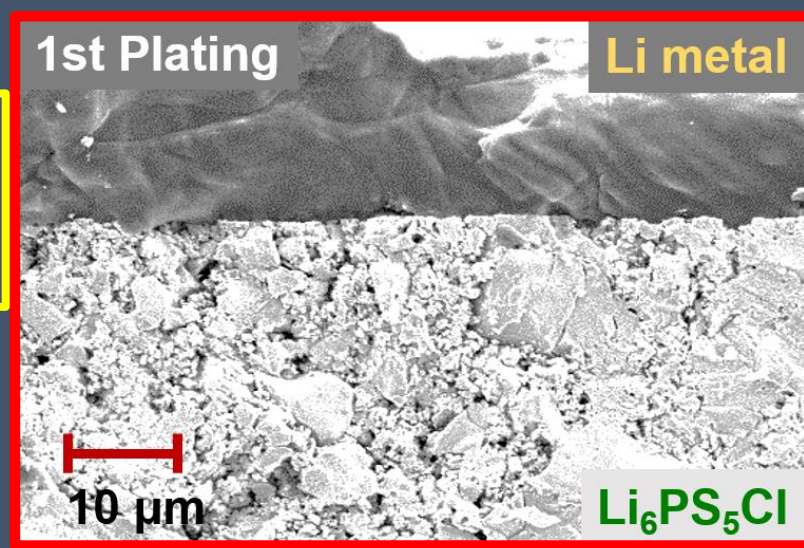
Void formation

Li₆PS₅Cl

SEM cross-sections

1st Plating

Li metal



10 μm

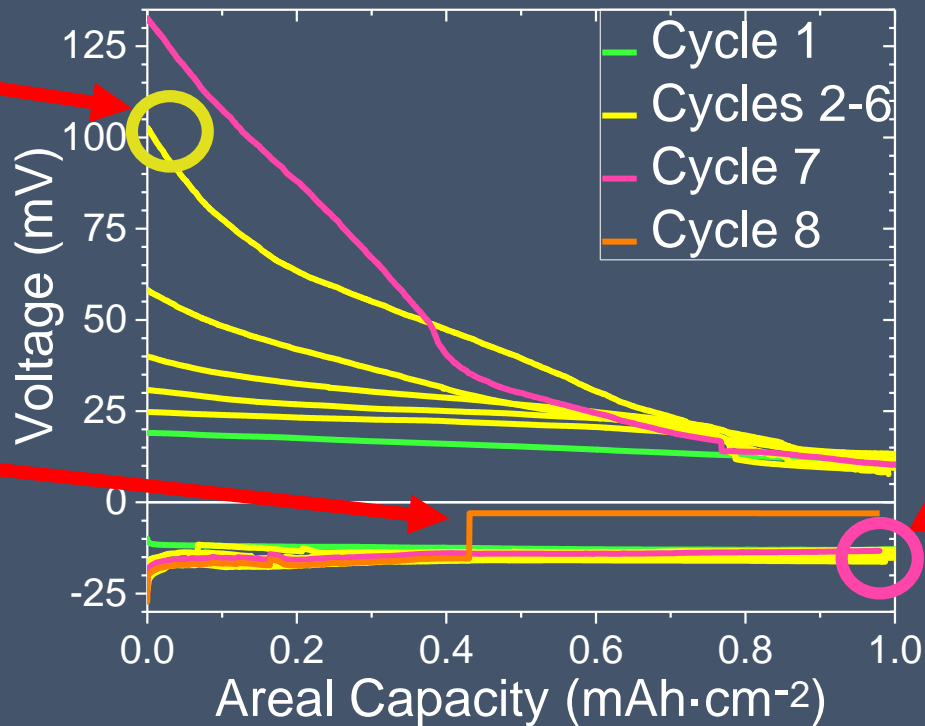
Li₆PS₅Cl

3-electrode cycling Li/Li₆PS₅Cl 3 MPa

Significant polarization increase

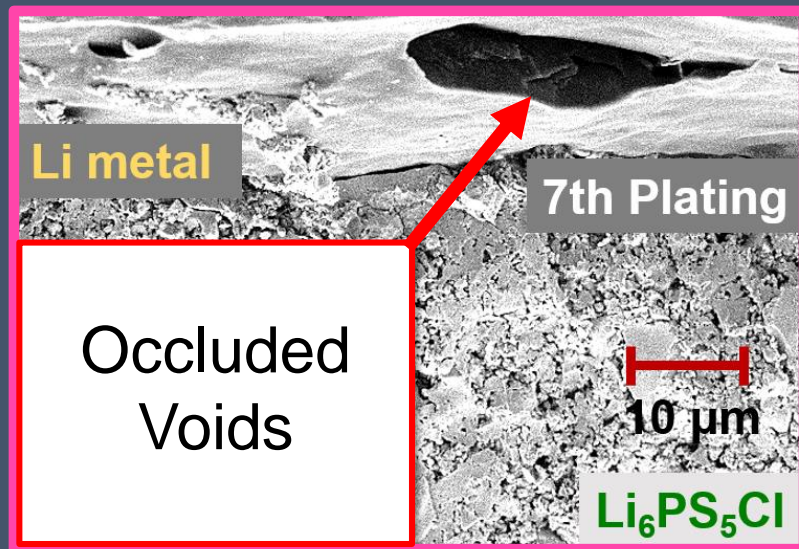
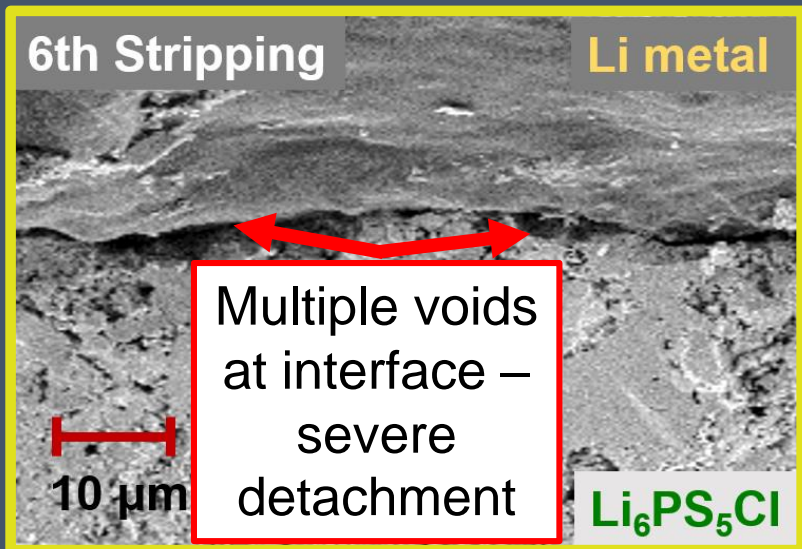
Eventual dendrite formation

short circuit



*Kasemchainan et al.
Nature Materials (2019)*

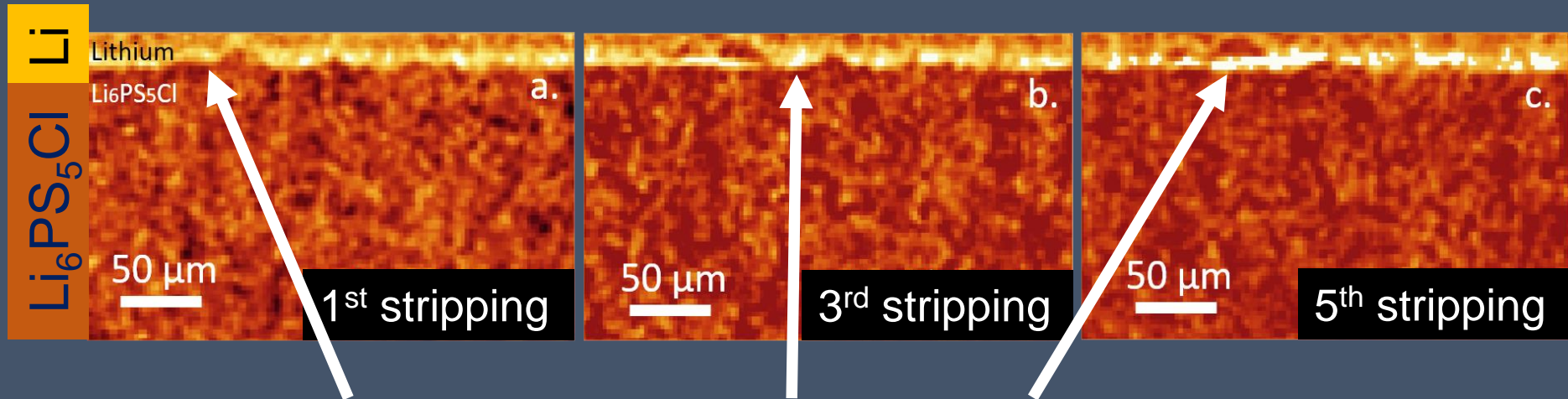
No polarization increase



In-situ X-ray computed tomography to reveal void evolution at the interface

Observe void evolution under controlled pressure cycling

Reconstructed virtual cross section from CT



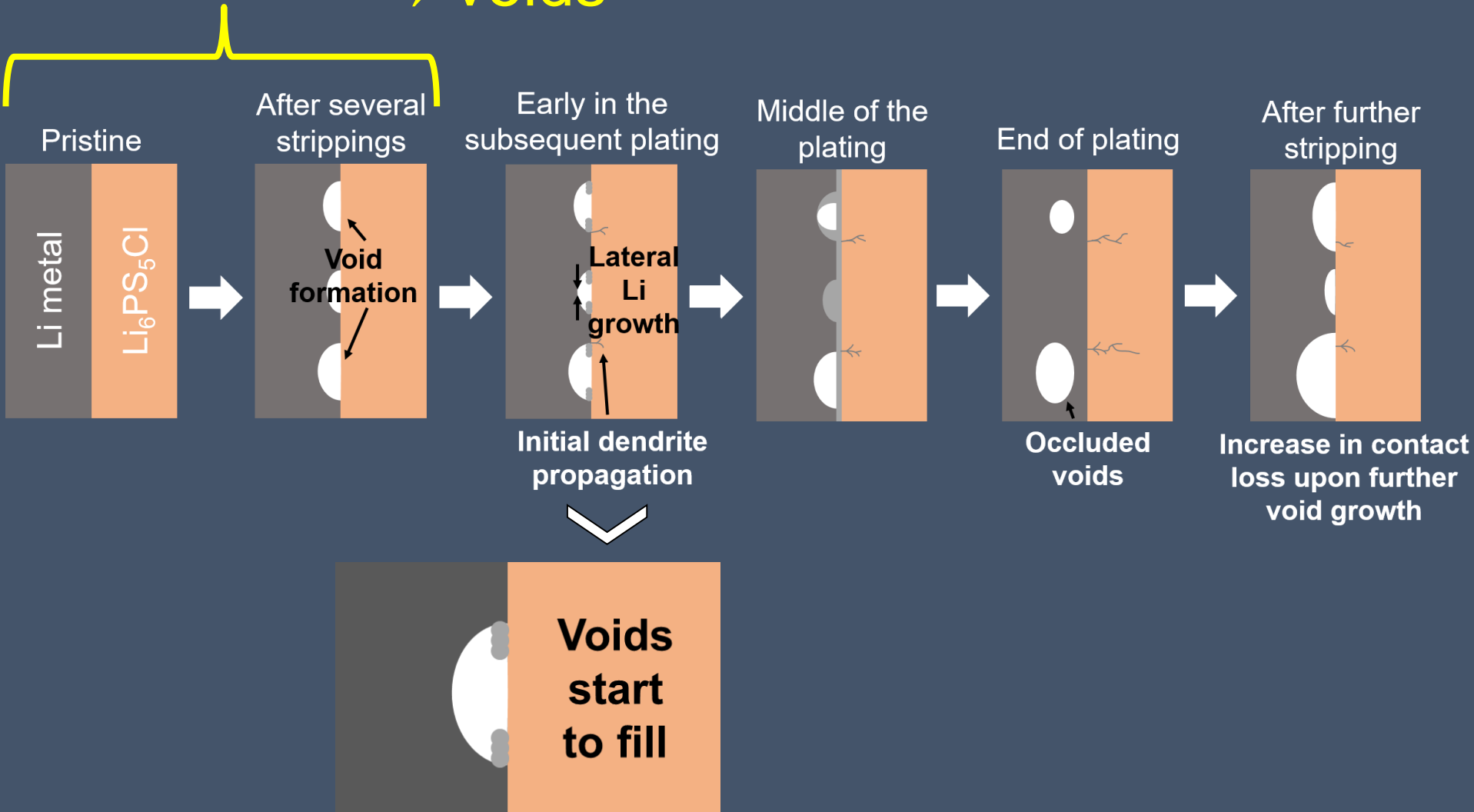
Void formation during cycling

Number and length of voids increase along interface

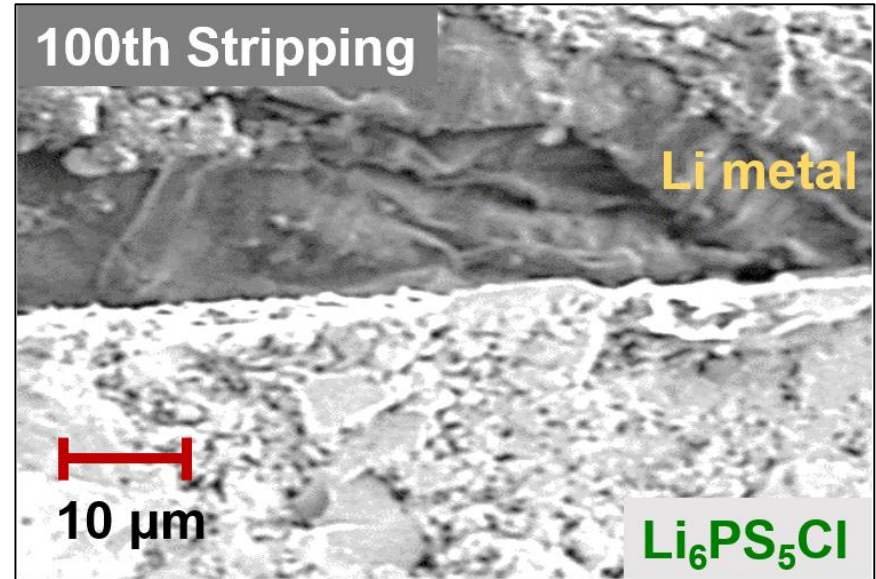
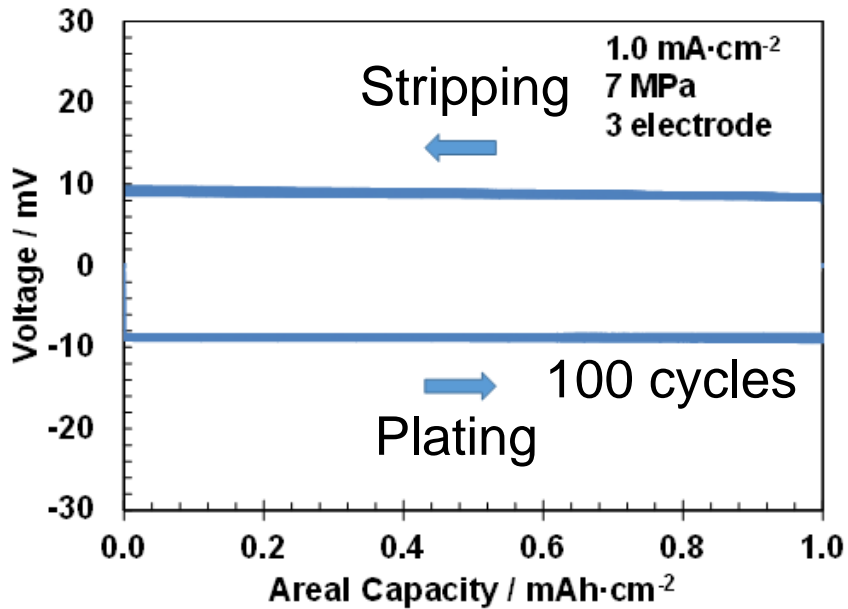
Li / Li₆PS₅Cl interface
Current density = 1.0 mA cm⁻²
Capacity = 1.0 mAh cm⁻²
Pressure = 3 MPa

Influence of void formation on cycling

$$J_{Li\text{ diffusion}} + J_{Li\text{ creep}} < J_{Li^+\text{ migration}} \rightarrow \text{voids}$$



3-electrode cycling Li/Li₆PS₅Cl 7 MPa



High pressure → Conformal interface

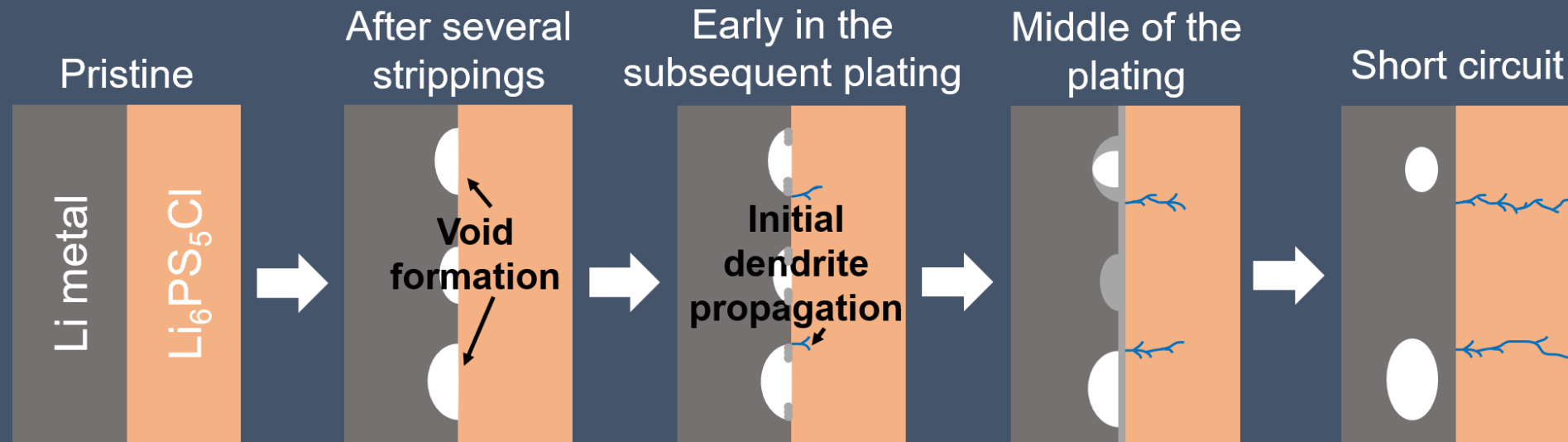
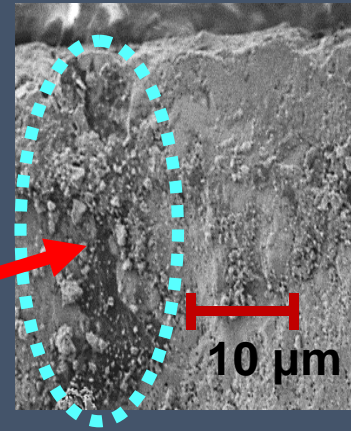
- Strong pressure dependence – creep of Li metal to interface
- Pressure \gg yield strength of Li: 0.81 MPa [1]

Implications for failure of solid state batteries

SSBs fail because of critical stripping current (CCS)
not plating current (CCP) as thought previously

Above CCS:

- Increase in voiding with cycle number – contact loss
- increase in local current density at interface
- reach critical current for **dendrite formation**
- dendrite propagating through solid electrolyte



Acknowledgements



Prof. Peter Bruce
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Dr. Stefanie Zekoll
Ziyang Ning



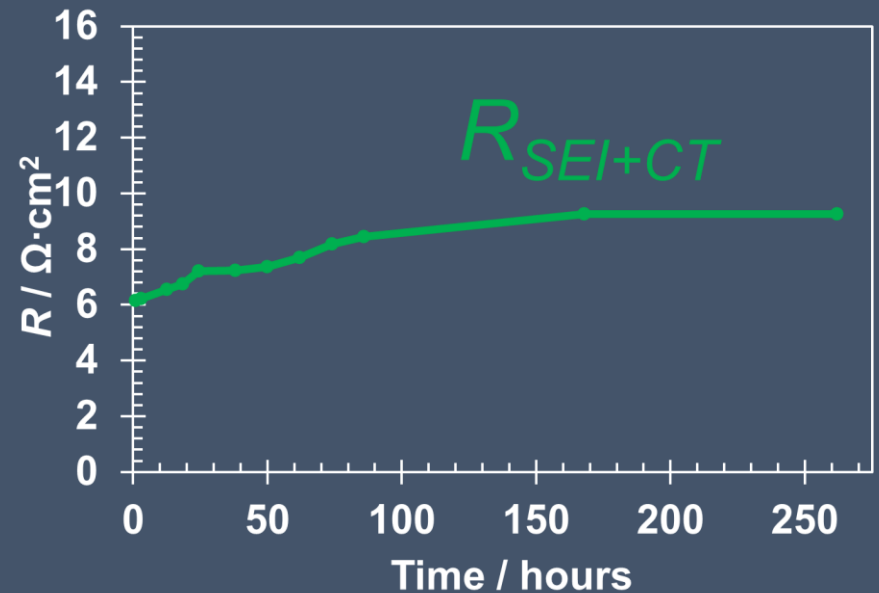
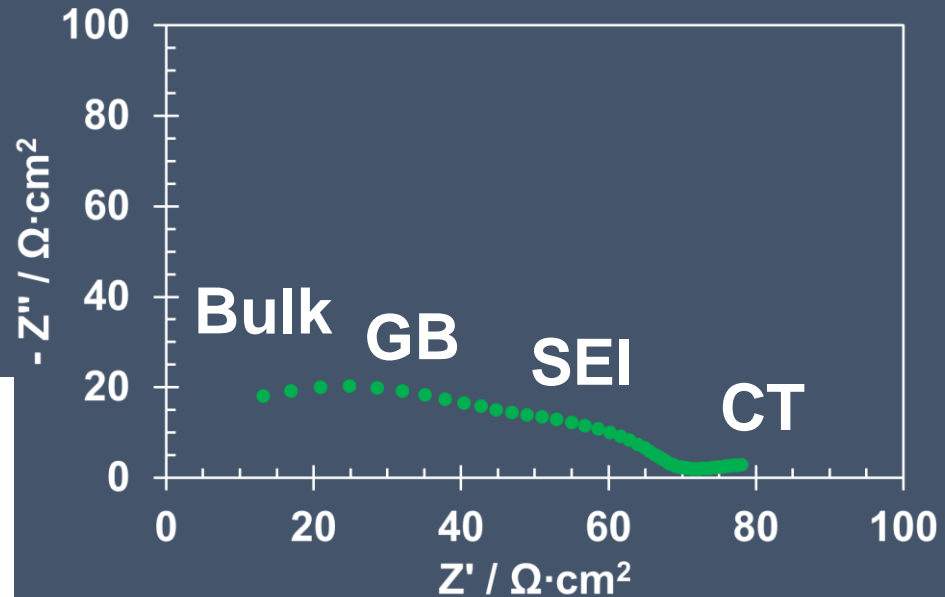
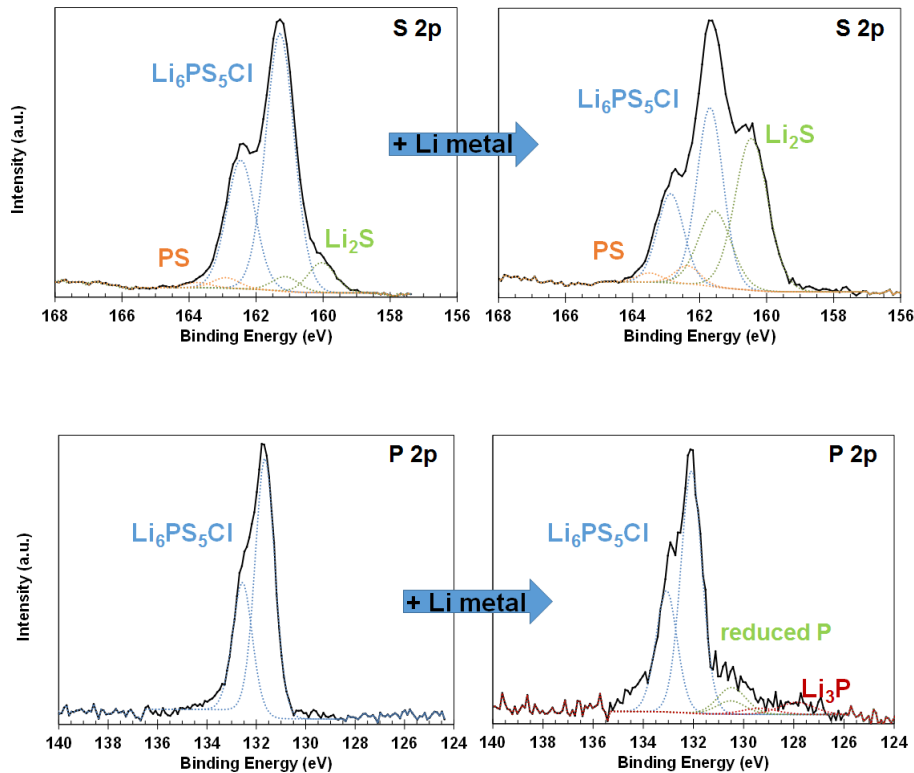
EPSRC (SUPERGEN)



ALiSTORE ERI

Li/Li₆PS₅Cl Interfacial Stability

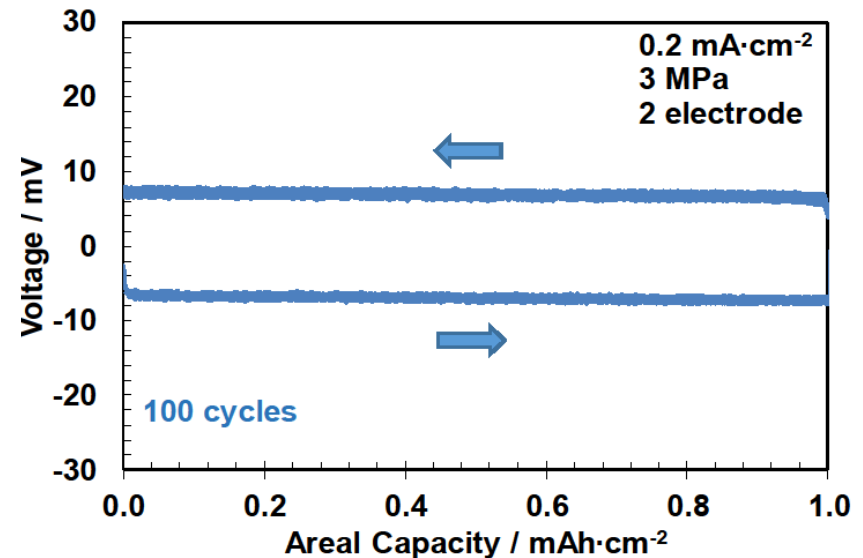
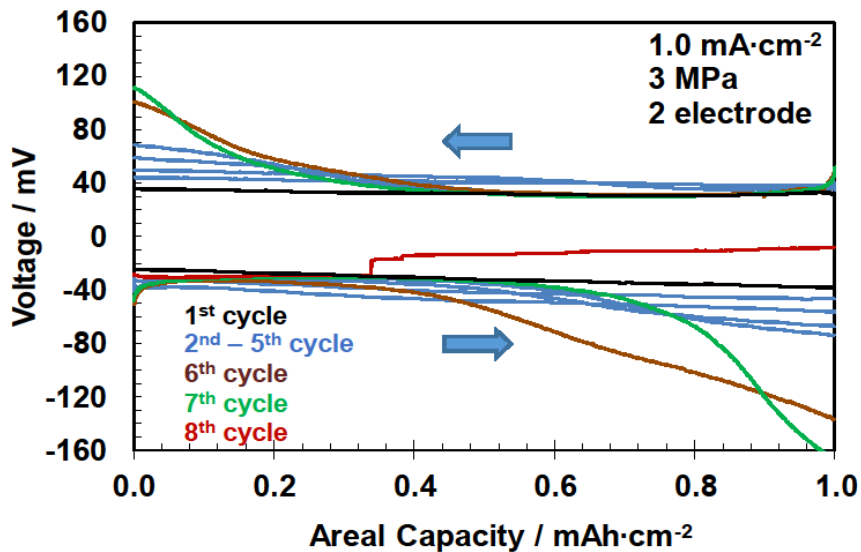
XPS shows that Li₆PS₅Cl is reduced by Li to form mainly Li₂S & Li₃P



2-electrode Li/Li₆PS₅Cl/Li cells

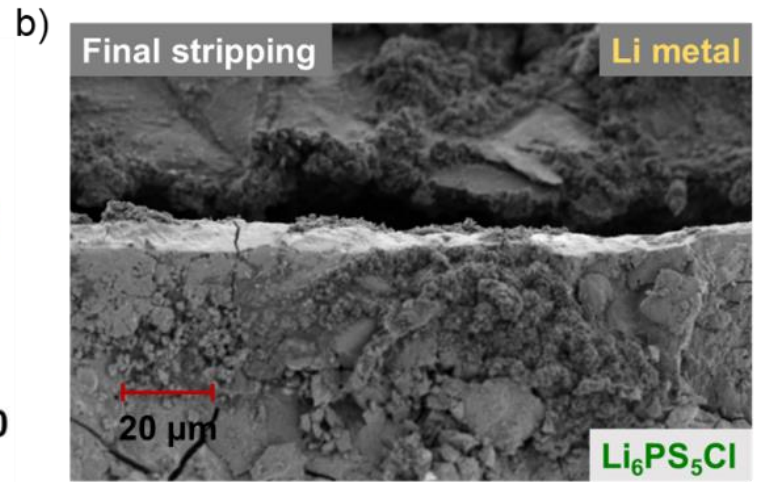
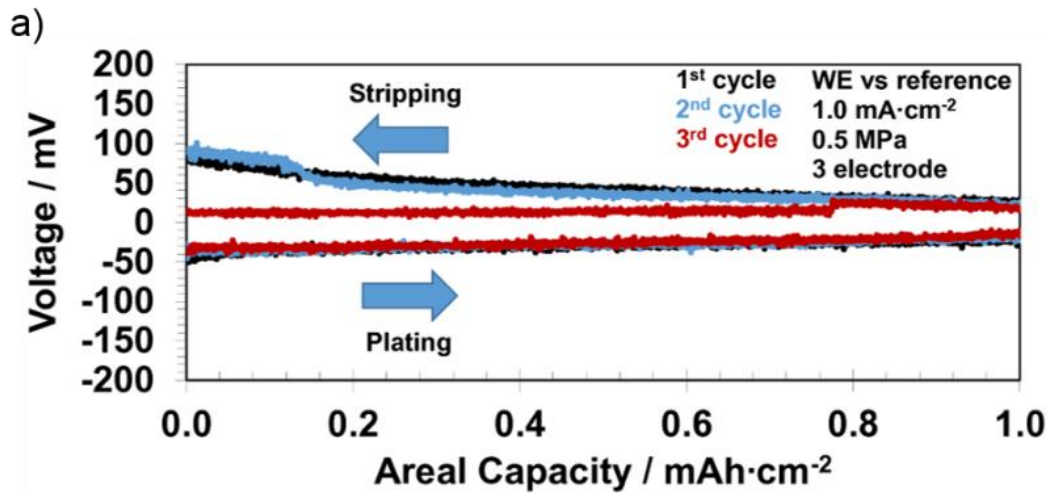
Current > critical stripping current
Current < critical plating current
cell failure

Current < critical stripping current



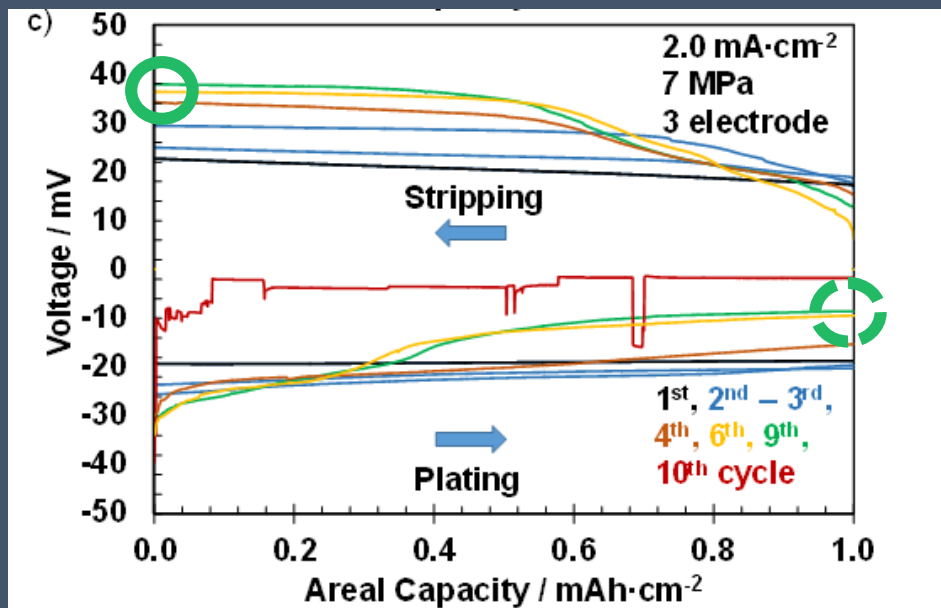
increasing polarization
followed by short circuit

$$J_{Li \text{ diffusion}} + J_{Li \text{ creep}} < J_{Li^+ \text{ migration}} \rightarrow \text{cell failure}$$

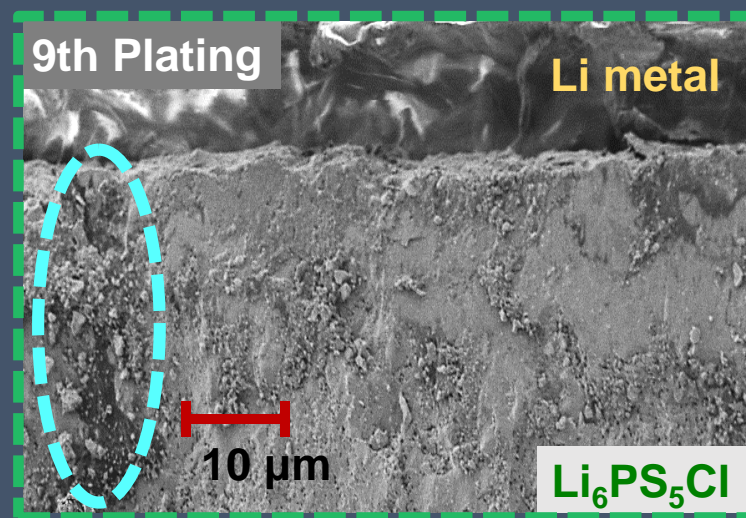
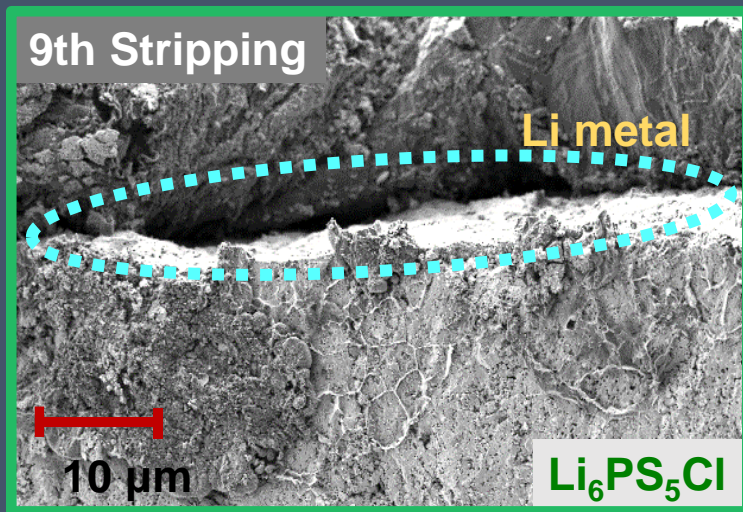


Voltage vs charge passed for a 3-electrode cell on Li metal plating and stripping at the Li / Li₆PS₅Cl interface at a pressure of 0.5 MPa and a current density of 1.0 mA·cm⁻² SEM cross-section of the Li / Li₆PS₅Cl interface after the final stripping.

3-electrode cycling Li/Li₆PS₅Cl 7 MPa 2 mAcm⁻²



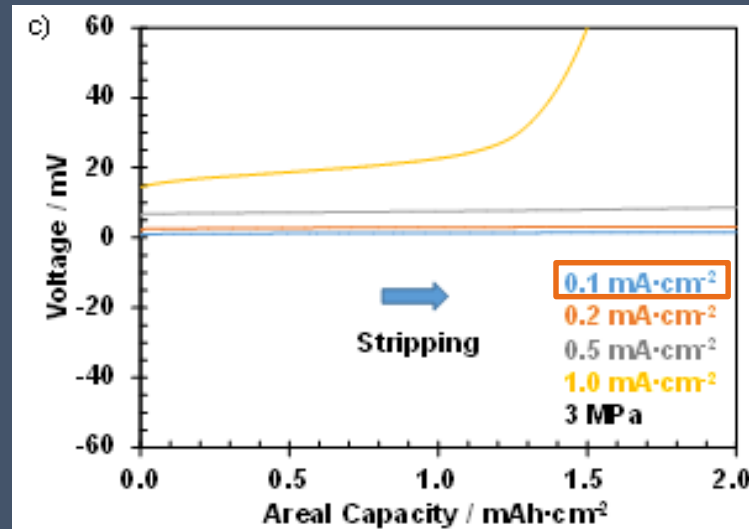
Reduced polarization
↓
dendrite formation



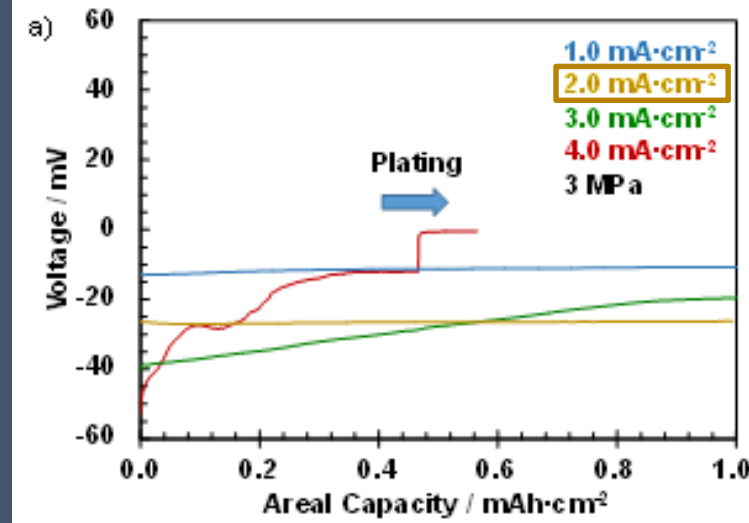
At 7 MPa and 2 mA cm⁻²: $J_{Li\ diffusion} + J_{Li\ creep} < J_{Li+ migration} \rightarrow$ cell failure

Critical current determination

Stripping



Plating



$$CCS < CCP$$